

FROM RESEARCH TO INDUSTRY

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LASER INDUCED BREAKDOWN
SPECTROSCOPY, A METHOD TO MEASURE
AND CHARACTERIZE THE PERSONAL
EXPOSURE TO AIRBORNE PARTICLES



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General information

- Context
- Project goal
- Principle

LIBS information

- Principle
- Interest

Results

- Qualitative analysis
- Quantitative analysis

Conclusion & Perspectives



Context

- Increase of utilization and production of nanomaterials in research and industry
- Discussions on potential health impacts caused by inhalation of particles

■ Risks?



- Current scarcity of hazard data in nanotoxicology
 - Particle capacity to reach and deposit in the deep alveolar regions of lungs
 - Exposure can be hazardous
 - Studies realized on cells or on animals but complicate to extrapolate to human
- To assess the efficiency of risk management measure, it is necessary to determine the personal exposure

Project goal

- Developing a global solution to evaluate the personal exposure to particles
- Validate a collection device based on electrostatic precipitation principle, coupled with on-line and off-line particles analysis

- User friendly
- Sampler (off-line analysis)
- Monitor (on-line analysis)
- In a broad range of size particles



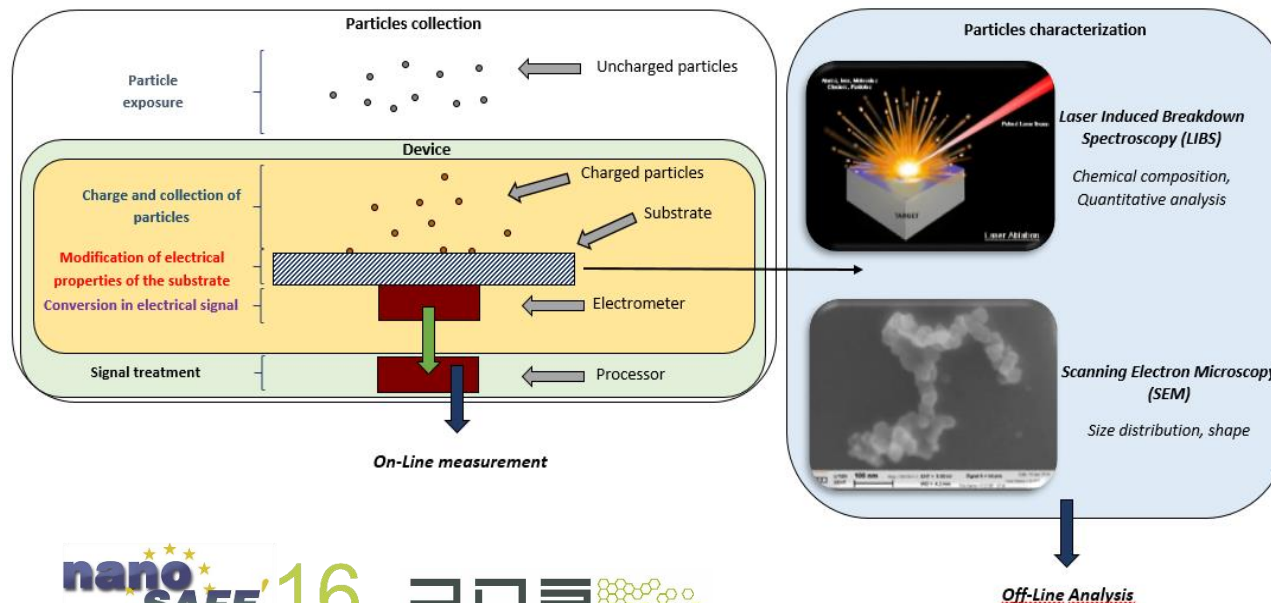
Electrostatic precipitator developed

- Wide scope
 - Environmental, Health and Safety issues
 - Worker protection
 - Inhalation toxicology

Principle

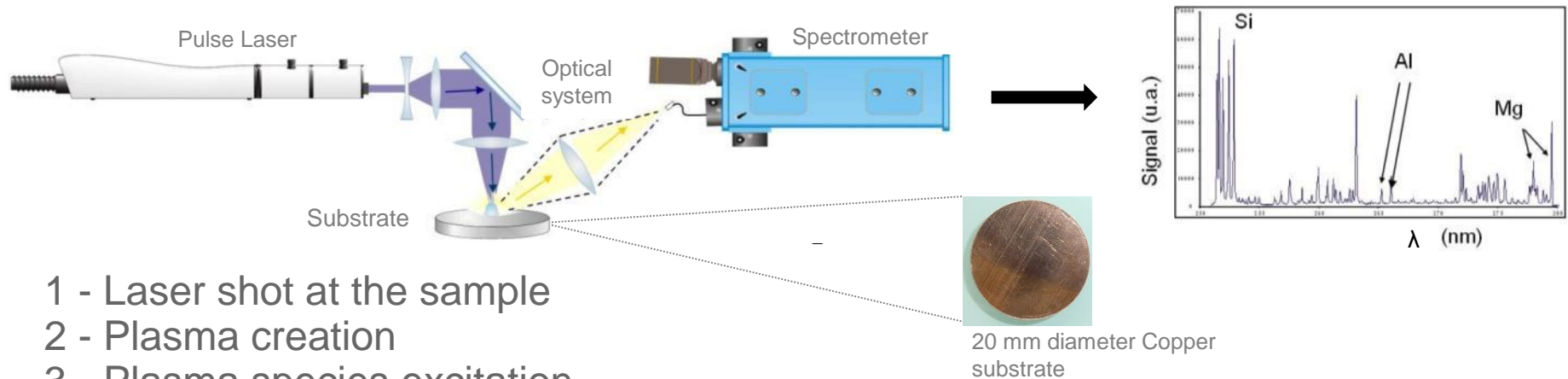
■ Five stages

- Airborne particles capture
- Particles charge by two mechanisms (field and diffusion charging)
- Particles collection on a metallic substrate by **size-dependent zone**
- On line measurement of concentration
- Off line analysis of chemical composition, spatially resolved by Laser Induced Breakdown Spectroscopy (LIBS)



More information about device functioning in PS2-9, 8th November

Principle



- 1 - Laser shot at the sample
- 2 - Plasma creation
- 3 - Plasma species excitation
- 4 - Radiation emissions, characteristic of chemical composition

Advantages

- No sample preparation
- Fast analysis
- Spatially resolved, one spot \approx 100 - 250 μ m

Disadvantages

- Reproducibility and accuracy discussed
- Destructive analysis

Interest of LIBS for this project :

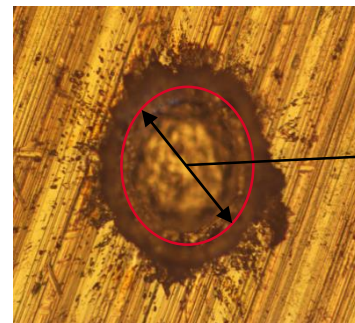
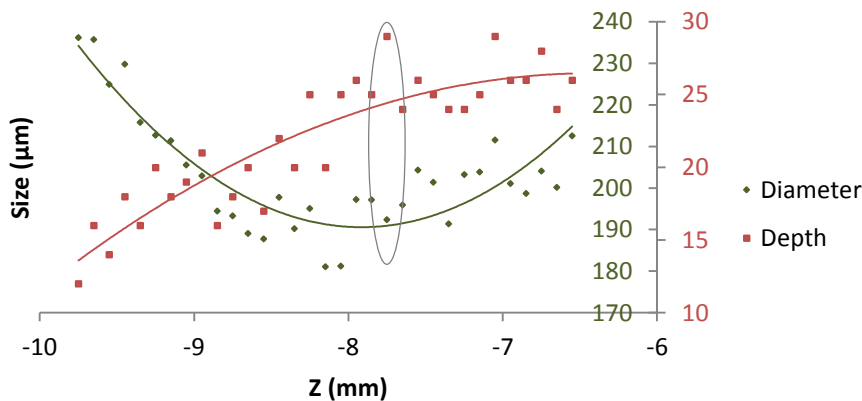
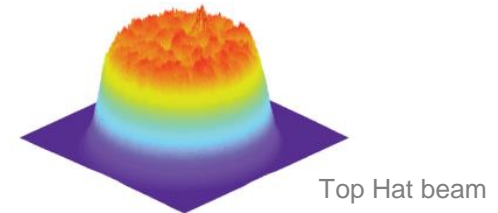
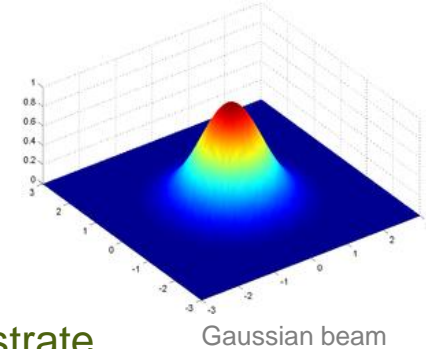
Spatial resolution

Analysis for each size-dependent zones at the substrate surface

Fast analysis

Laser beam adjustment

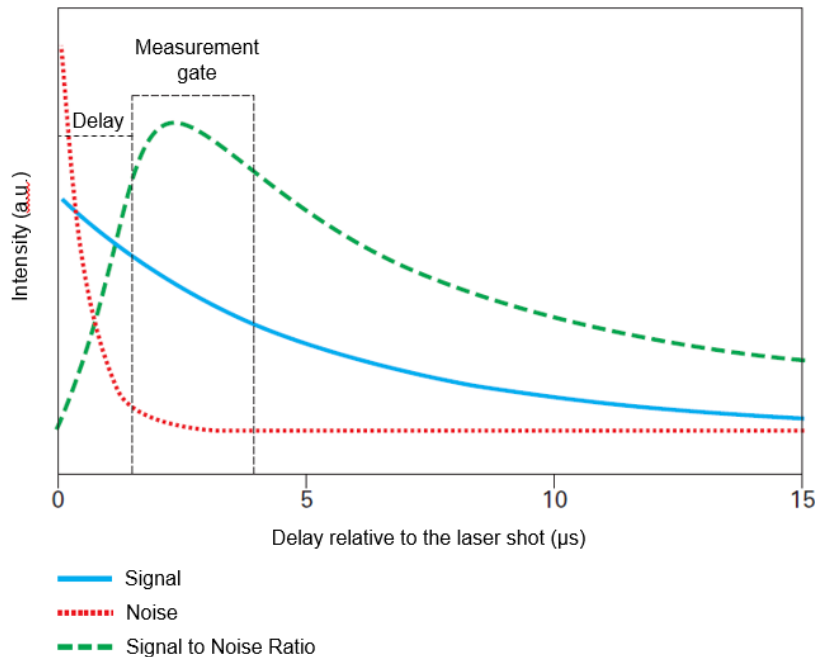
- Beam created by laser is a Gaussian beam
- Top Hat beam : Homogeneous beam
 - Important to create a homogeneous crater
 - Important to provide the same energy at any point of the substrate
- Necessity to adjustment of optical system
 - Distance sample – mirror (Z)
 - Diameter and depth measurement by microscopy
 - Max depth for min diameter



Typical crater

Spatial resolution
Diameter : 192 μm
Depth : 29 μm

Signal optimization – SNR improvement

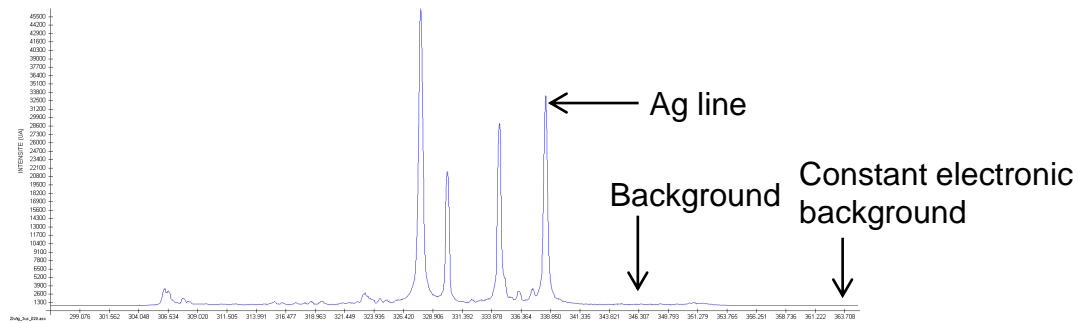


- Delay optimization
 - Delay : time between the end of laser shot and the signal acquisition
- $50 \text{ ns} \leq \text{Delay} < \text{some } \mu\text{s}$
- If Delay \uparrow , signal intensity \downarrow

- First selection of five metallic materials for substrate – Cu, Ti, Al, Zn, Ni
- Selection of five particulate materials for deposit – TiO_2 , ZnO, SiO_2 , Ag, Al_2O_3
- Choice of 16 couples "substrate / particles" without spectral interferences from databases

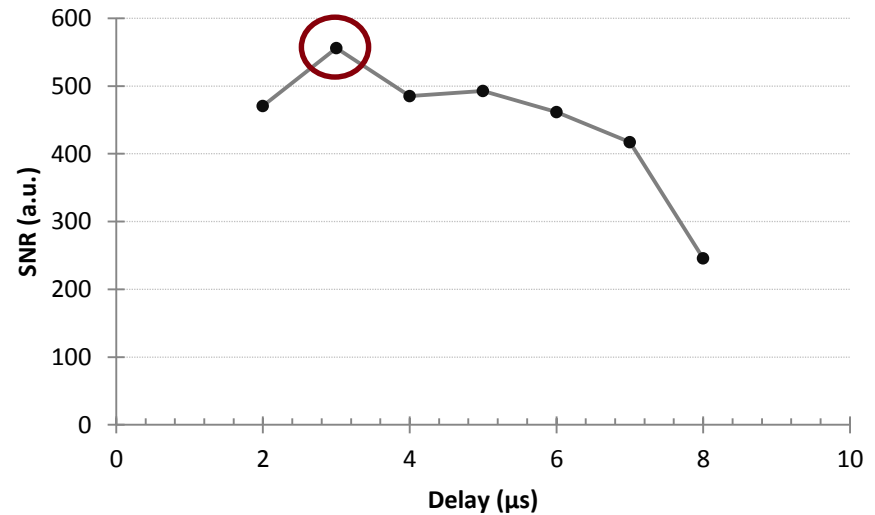
Signal optimization – SNR improvement

Ag deposit on Zn substrate



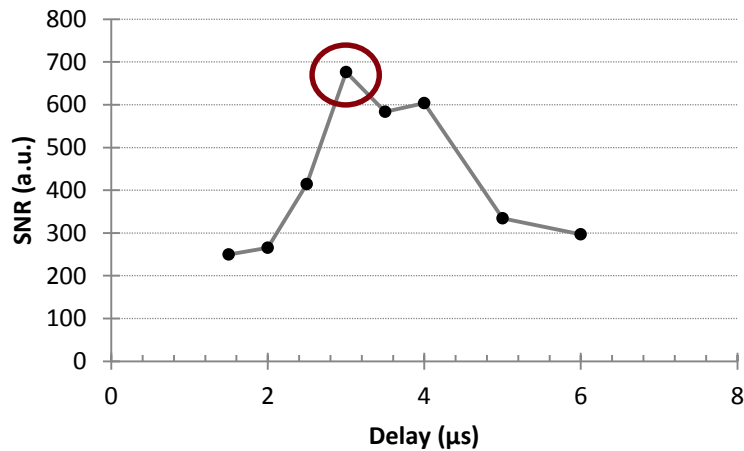
- Ag : 338.3 nm
- Background : 346.7 nm
- Electronic background 825 a.u.

- Optimal delay : 3 μ s
- SNR max : 555 a.u.



Signal optimization – SNR improvement

Ag deposit on Cu substrate

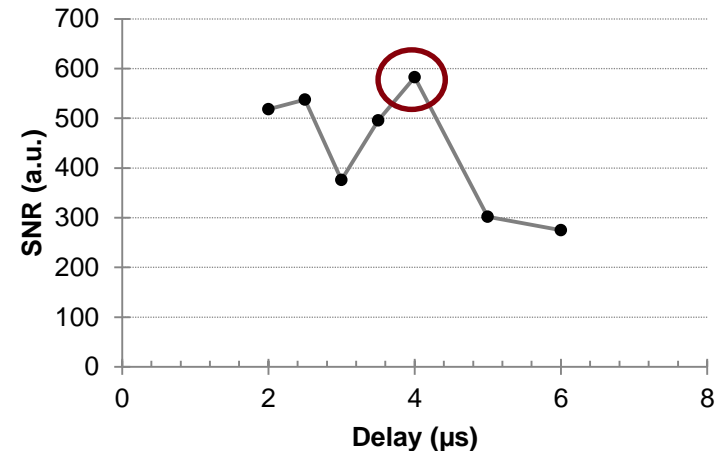


■ Ag : 338.3 nm

■ Optimal delay : 3 μs

■ SNR max : 676 a.u.

Ag deposit on Al substrate



■ Ag : 338.3 nm

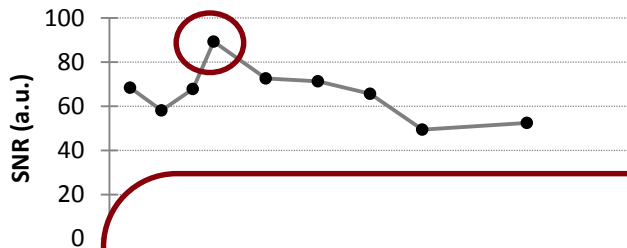
■ Optimal delay : 4 μs

■ SNR max : 582 a.u.

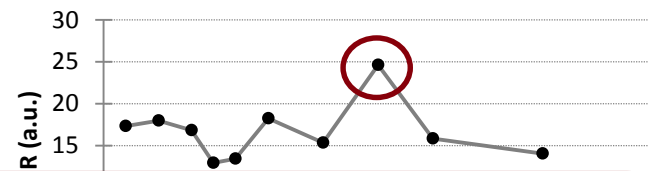
→ Optimal delay for Ag deposit : 3 - 4 μs, for three different substrates

Signal optimization – SNR improvement

SiO₂ deposit on Cu substrate



SiO₂ deposit on Ni substrate

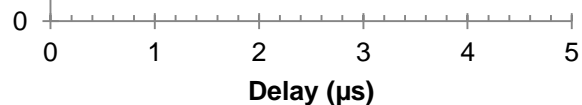


Determination of optimal delay necessary for each new sample (deposit / substrate)

Optimal delay reusable for same samples (deposit / substrate)

Optimal delay
SNR max

2.5 μs
a.u.



Optimal delay : 3 μs
SNR max : 205 a.u.

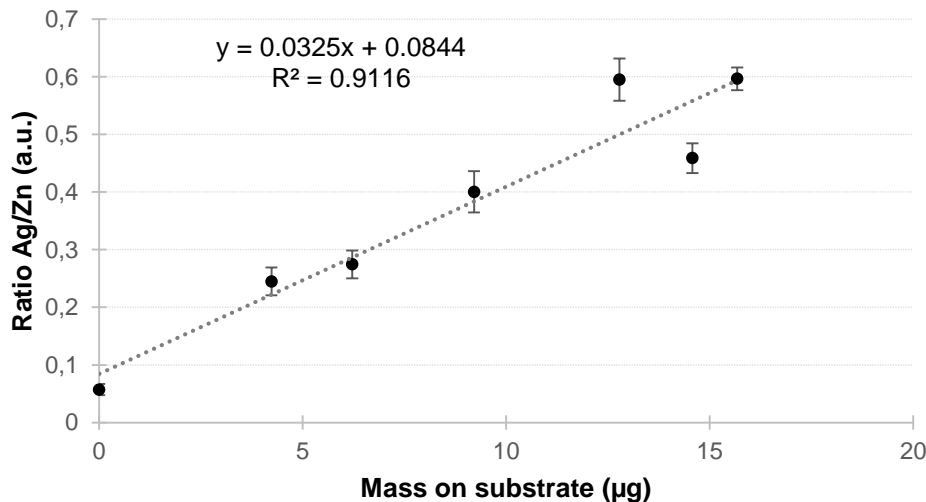
→ SiO₂ three different substrates / three different optimal delays

Calibration – Ag example

- Realization of controlled deposits (0.25, 0.5, 1, 2, 2.5 & 3 hours)
 - Ag particles
 - On Cu substrate and Zn substrate
- First quantification of deposit by XRF analysis using existing calibration
- Calibration by LIBS
 - Normalization
 - Determination $I_{\text{Ag}}/I_{\text{Substrate}}$
- Determination of LOD and LOQ
 - LOD $3\sigma/a$
 - LOQ $10\sigma/a$

Calibration – Ag example

Ag deposit on Zn substrate



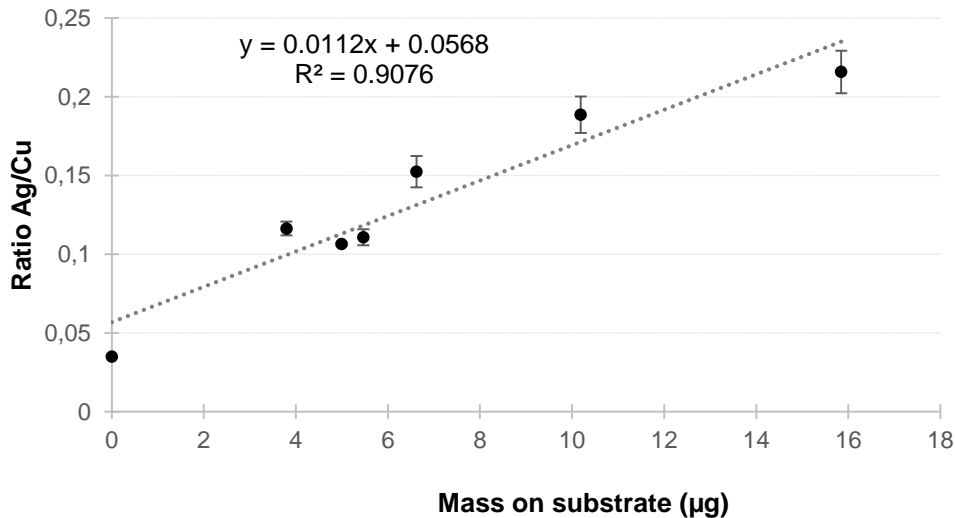
- LOD $3\sigma/a = 1.8 \mu\text{g}$
- LOQ $10\sigma/a = 6 \mu\text{g}$
- Substrate area = 3.14 cm^2
- LOD = $0.57 \mu\text{g}/\text{cm}^2$
- LOQ = $1.9 \mu\text{g}/\text{cm}^2$

Estimation of airborne concentration :

During 8h sampling at 1L/min, it is possible to detect a particle aerosol of $3.9 \mu\text{g}/\text{m}^3$ and to quantify an aerosol of $12.5 \mu\text{g}/\text{m}^3$

Calibration – Ag example

Ag deposit on Cu substrate

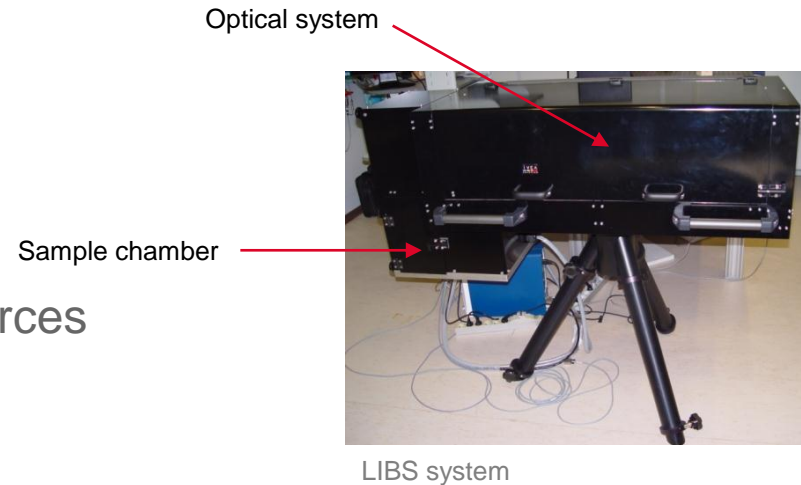


- LOD $3\sigma/a = 0.28 \mu\text{g}$
- LOQ $10\sigma/a = 0.93 \mu\text{g}$
- Substrate area = 3.14 cm^2
- LOD = $0.089 \mu\text{g}/\text{cm}^2$
- LOQ = $0.298 \mu\text{g}/\text{cm}^2$

Estimation of airborne concentration :

During 8h sampling at 1L/min, it is possible to detect a particle aerosol of $0.6 \mu\text{g}/\text{m}^3$ and to quantify an aerosol of $1.9 \mu\text{g}/\text{m}^3$

- Determination of the spatial resolution
 - Top Hat beam
 - $\approx 200 \mu\text{m}$
- Identification of the different noise sources
 - Electronic noise (825 a.u.)
 - Shot noise (below 1500 a.u.)
- SNR optimization by the delay modification
 - Convergence for some samples but...
 - For measurement quality, necessity to do this optimization for each new sample
- Studies of the shot energy, the substrate materials, the substrate rugosity, ...
 - Determination of reference material for substrate (Cu or Zn)
- Calibration and determination of LOD and LOQ
 - $\approx \mu\text{g}/\text{m}^3$

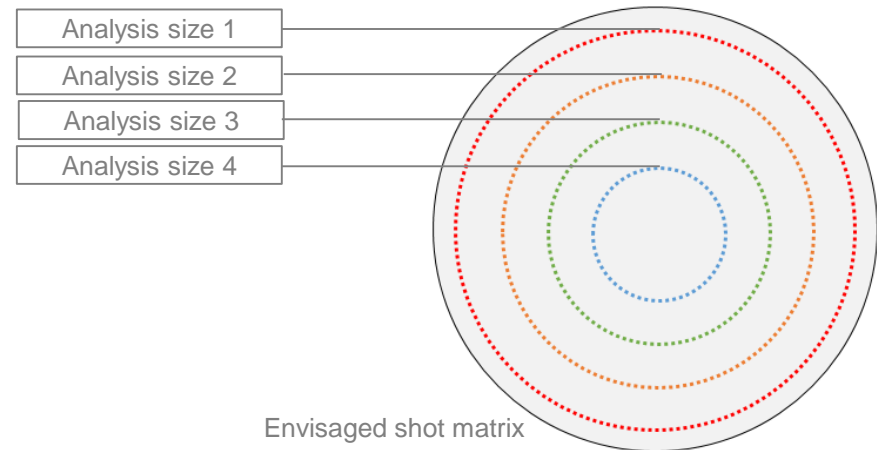


- Final precipitator permits to collect particles along concentric circles with a size dependent radius
 - A radius = A particle size

- With LIBS, possibility to shot along concentric circles
 - A LIBS analysis by radius, so by size
 - Quantification and chemical composition determination for each size

- Provide a global analysis of the deposit
 - Size distribution
 - Chemical composition

- Extrapolation to the aerosol





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**Thank you for
your attention !**



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