

# Effect of carbon black nanoparticles on methane/air explosions: Influence at low initial turbulence

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David Torrado, Pierre-Alexandre Glaude & Olivier Dufaud

Laboratoire Réactions et Génie des Procédés

LRGP - Nancy



# CONTEXT

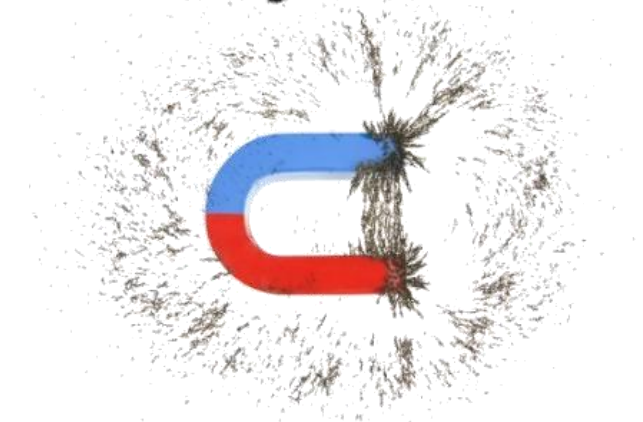
## Nanoparticles used to modify:

- Resistance
- Surface modification
- Rheology
- Magnetization

## High Specific Surface

→ Advantageous on :

- Catalysis
- Biomaterials
- Biological applications



**Necessity of studies on the toxicity and explosion hazards of combustible nanoparticles**

# EXPLOSION OF NANOPARTICLES

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## Previous works have shown:

- The ignition sensitivity can increase for nano-dusts
- Similar explosion severity for  $\mu\text{m}$ - and nanoparticles
  - Dispersion of a highly loaded cloud → Utopic
  - Fast agglomeration / enhancing sedimentation

**Unlikely to occur at normal industrial conditions**

## **However...** Simultaneous gas/vapor and dispersed nanoparticles

→ Hybrid mixture explosion

- ① Could the **explosion be modified** by the presence of nano-dust at **low turbulences levels**?
- ② Could an initial concentration of **carbon black nanoparticles increase the severity** of a **hybrid mixture explosion**?

# PRESENTATION CONTENTS

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Materials & Methods

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graph TD; A[Materials & Methods] --> B[Results]; B --> C[Conclusions]; B --- B1[① DISPERSION STABILITY]; B --- B2[② EXPLOSION SEVERITY – Influence of dust concentration]; B --- B3[③ EXPLOSION SEVERITY – Influence of turbulence];
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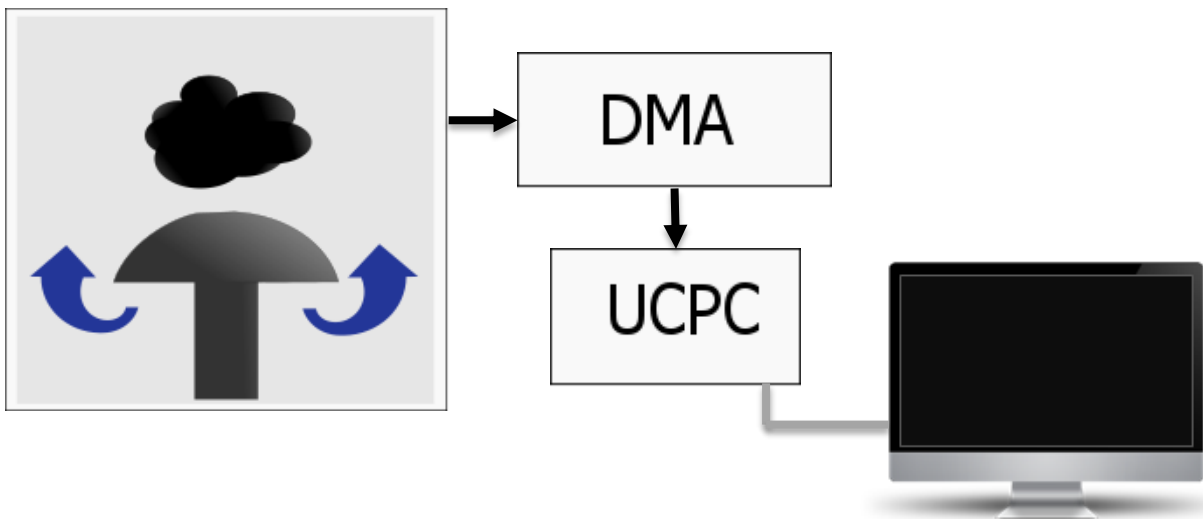
Results

- ① DISPERSION STABILITY
- ② EXPLOSION SEVERITY –  
Influence of dust concentration
- ③ EXPLOSION SEVERITY –  
Influence of turbulence

Conclusions

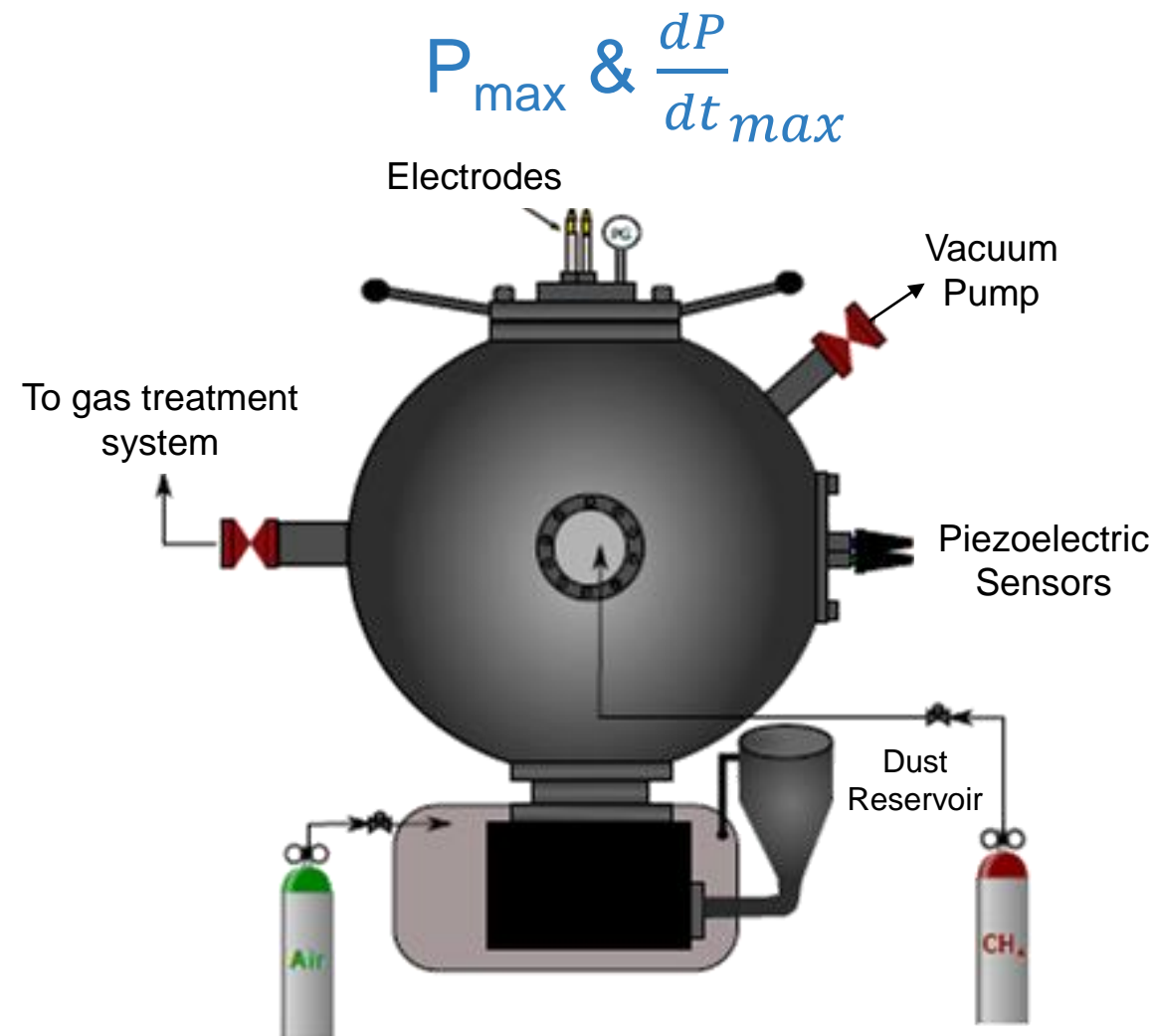
# EXPERIMENTAL SETUP

## Dispersion Stability



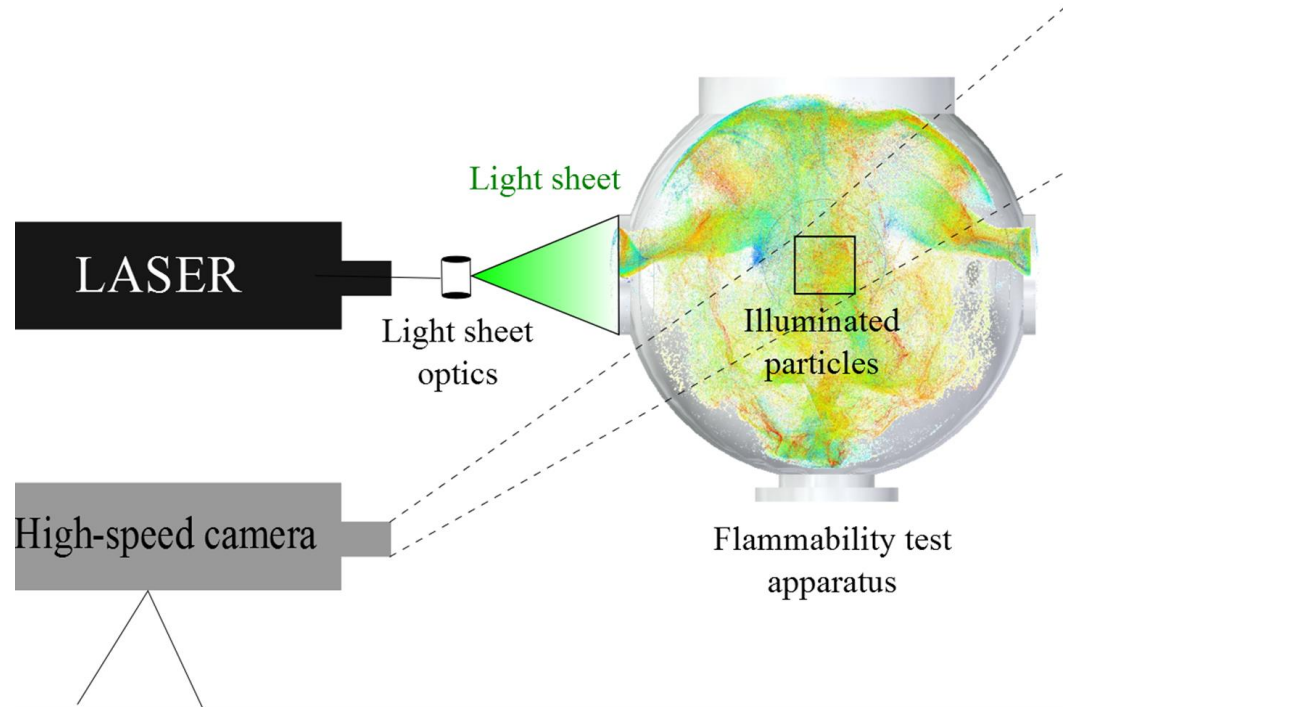
### Tested by:

- Differential Mobility Analyzer (DMA)
- Ultrafine Condensation Particle Counter (UCPC)



Burnt gases were analysed by  $\mu$ GC

# IGNITION DELAY TIME - TURBULENCE



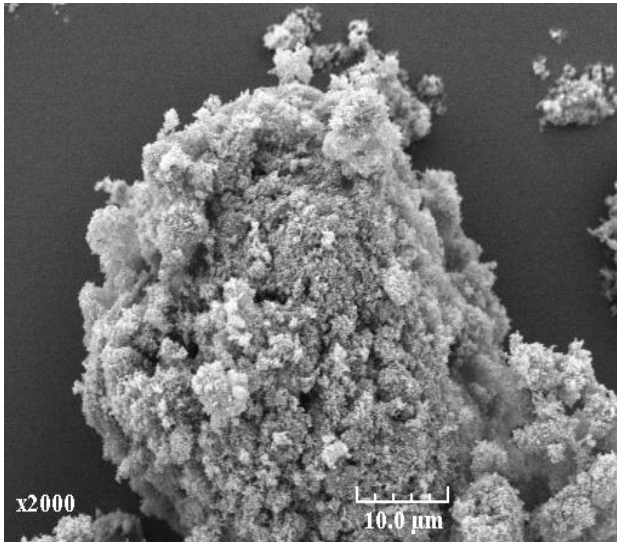
Particle Image Velocimetry (PIV)

Standard Value

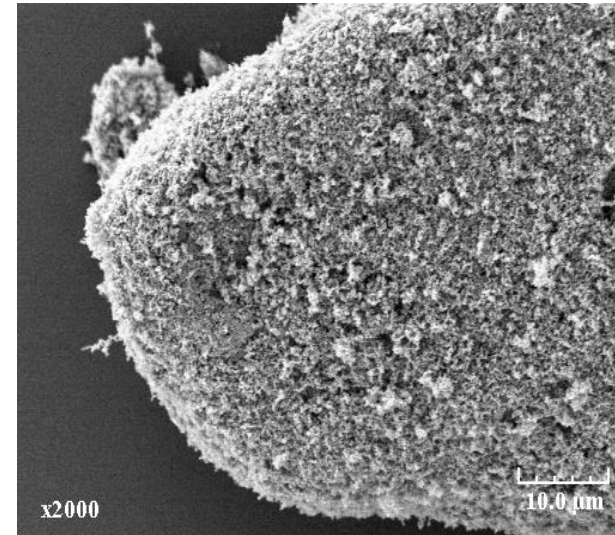
Ignition delay $t_v$ (ms)	Intensity of velocity fluctuations $u'$ (m.s <sup>-1</sup> )	
	Sphere (this study)	Sphere (Dahoe 2000)
<b>0</b>	0	0
<b>60</b>	3.40	3
<b>120</b>	1.05	0.9



# CARBON BLACK PARTICLES



**PRINTEX XE2**



**CORAX N550**

Ignition delay $t_v$ (ms)	Powder concentration (g/m <sup>3</sup> )		
	0	0.5	2.5
0	Gas Concentration (%v.): 5.5 - 7 - 8 - 9 - 12		
60			
120			

**Minimum Explosible Concentration (Carbon Black): 60 g/m<sup>3</sup>**

# PRESENTATION CONTENTS

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Materials & Methods



Results

- ① DISPERSION STABILITY
- ② EXPLOSION SEVERITY –  
Influence of dust concentration
- ③ EXPLOSION SEVERITY –  
Influence of turbulence



Conclusions

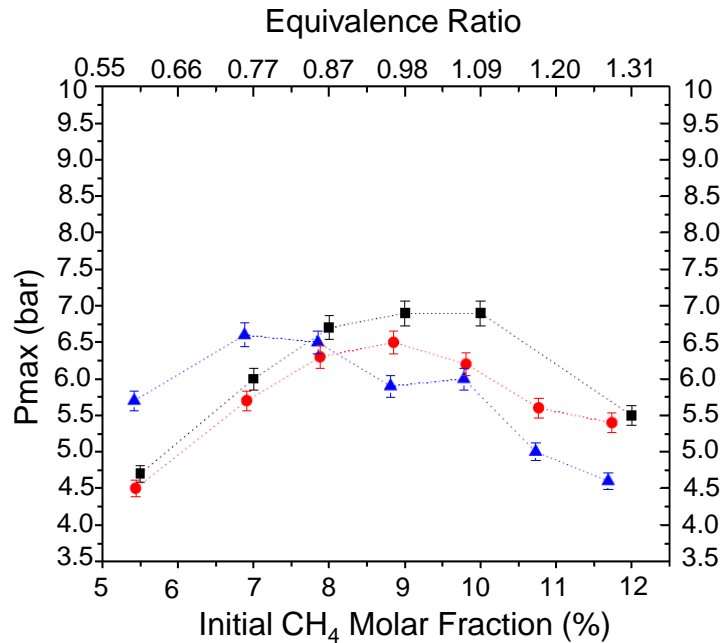


# DISPERSION STABILITY

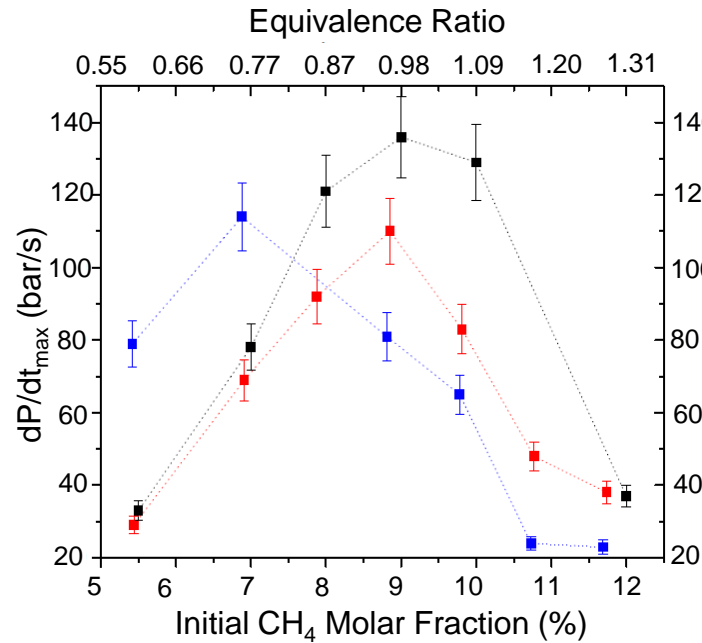
Nanopowders	Time after dispersion (s)	Mean Mobility Diameter (nm)	Concentration (particles/cm <sup>3</sup> )	Ambiance Concentration (particles/cm <sup>3</sup> )
Printex XE2	20	245	2.1x10 <sup>5</sup>	6x10 <sup>3</sup>
	190	241	1.5x10 <sup>5</sup>	
Corax N550	20	346	2.4x10 <sup>5</sup>	
	180	350	1.5x10 <sup>5</sup>	

- ① Similar **mobility diameter** after 20 and 190 seconds
- ② **Agglomeration phenomena** is limited because of the **low powder concentration**
- ③ A **stable dust cloud** of agglomerates is generated  
**Ignition delay times** ( $t_v$ ) → Present @ 60 & 120 ms

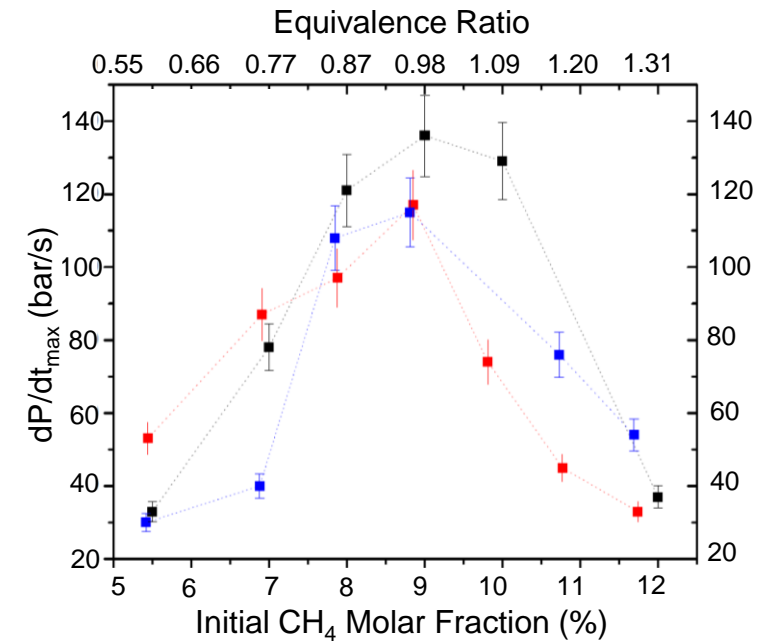
# INFLUENCE OF CB CONCENTRATION



Methane/Printex XE2/Air



Methane/Printex XE2/Air



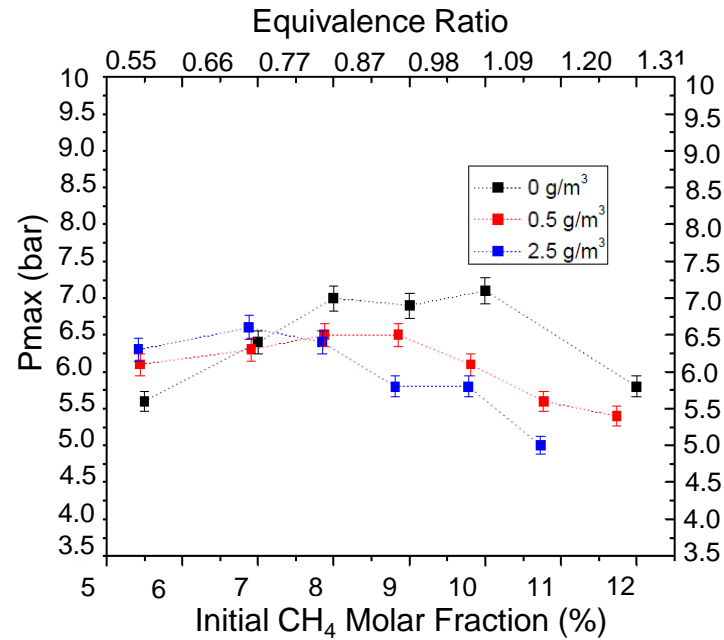
Methane/Corax N550/Air

■ 0 g.m<sup>-3</sup>    ● 0.5 g.m<sup>-3</sup>    ▲ 2.5 g.m<sup>-3</sup>

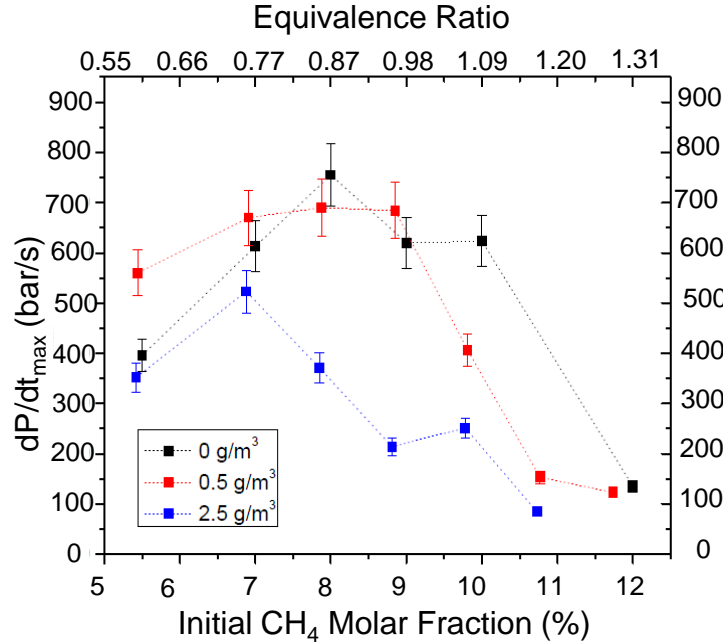
**At quiescent conditions (not dispersed at high pressure):**

- ① Higher explosions severity for initial CH<sub>4</sub> fractions 5.5-8%  
→ Same trends for Printex XE2 and Corax N550
- ② Strong influence of Printex XE2 on the explosion severity

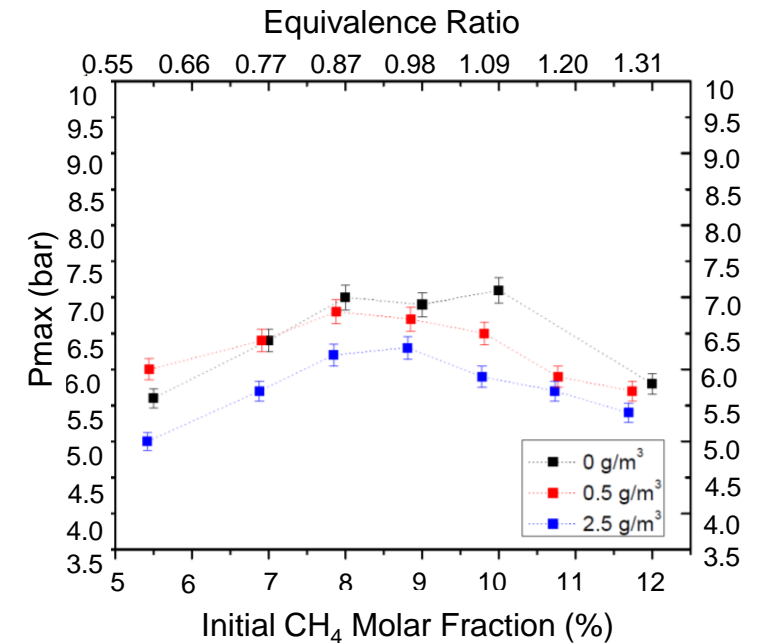
# INFLUENCE OF CB CONCENTRATION



Methane/Printex XE2/Air



Methane/Printex XE2/Air



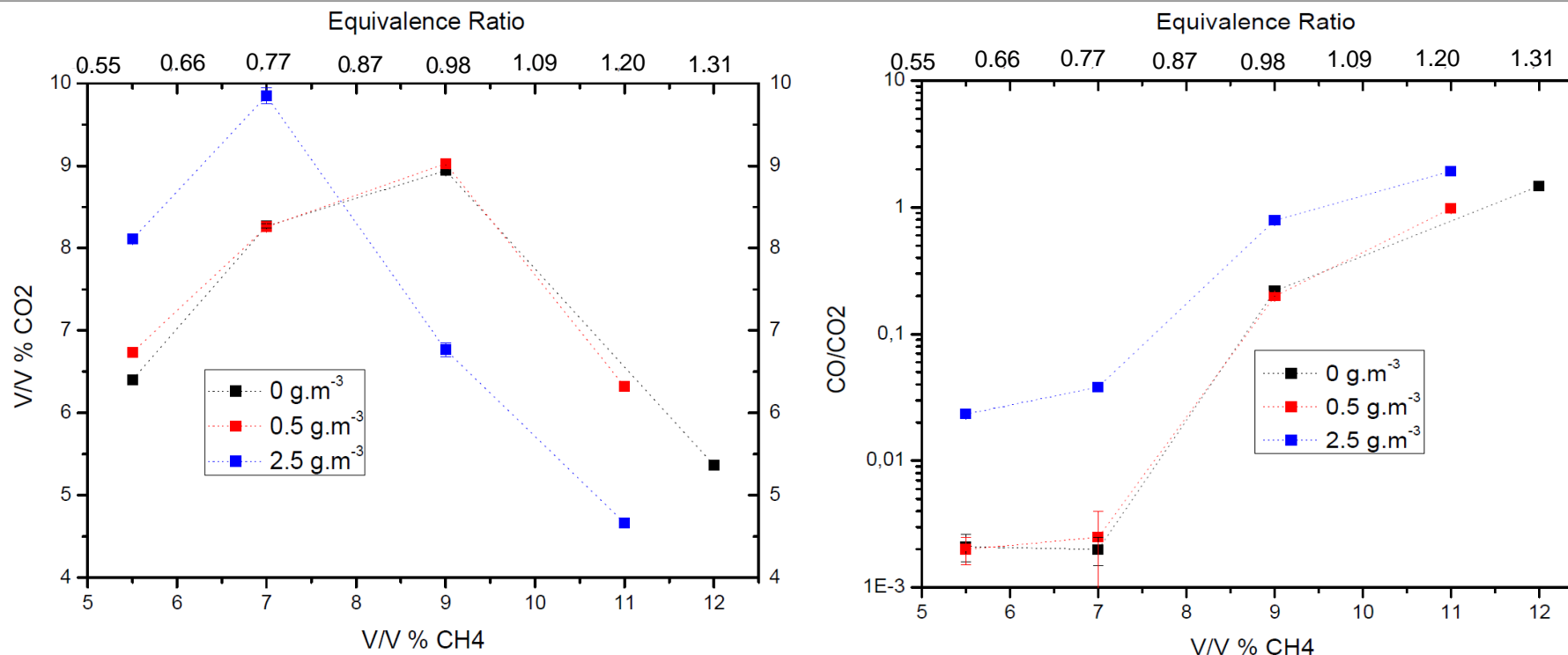
Methane/Corax N550/Air

0 g.m<sup>-3</sup>
 0.5 g.m<sup>-3</sup>
 2.5 g.m<sup>-3</sup>

At  $u' = 1.04 \text{ m/s}$  ( $t_v$  120ms):

- ① Similar trends compared to the quiescent system  
Higher influence at higher specific surface  $\rightarrow$  Deagglomeration of powder
- ② Turbulent quenching for Printex XE2  $\rightarrow$  No explosion at 12% v/v CH<sub>4</sub>

# INFLUENCE OF CB CONCENTRATION



*Printex XE2/Methane/Air mixture under quiescent conditions.*

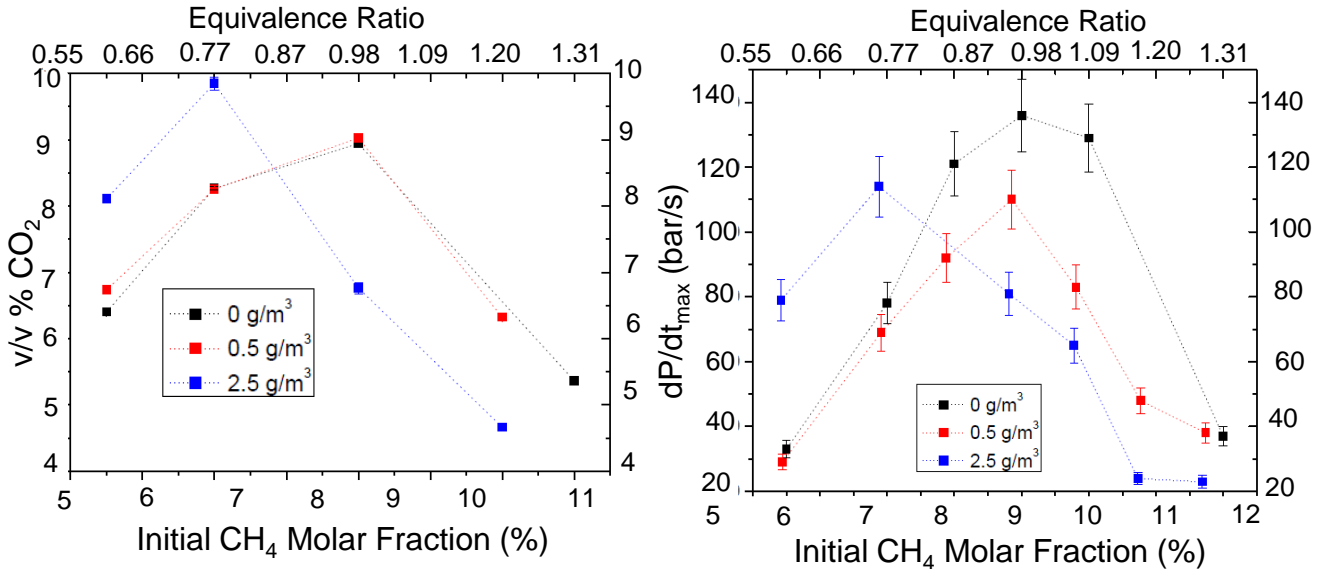
**Dispersion of a cloud** of carbon black affects the **combustion reaction**

↳ Even at **low concentrations of dust**

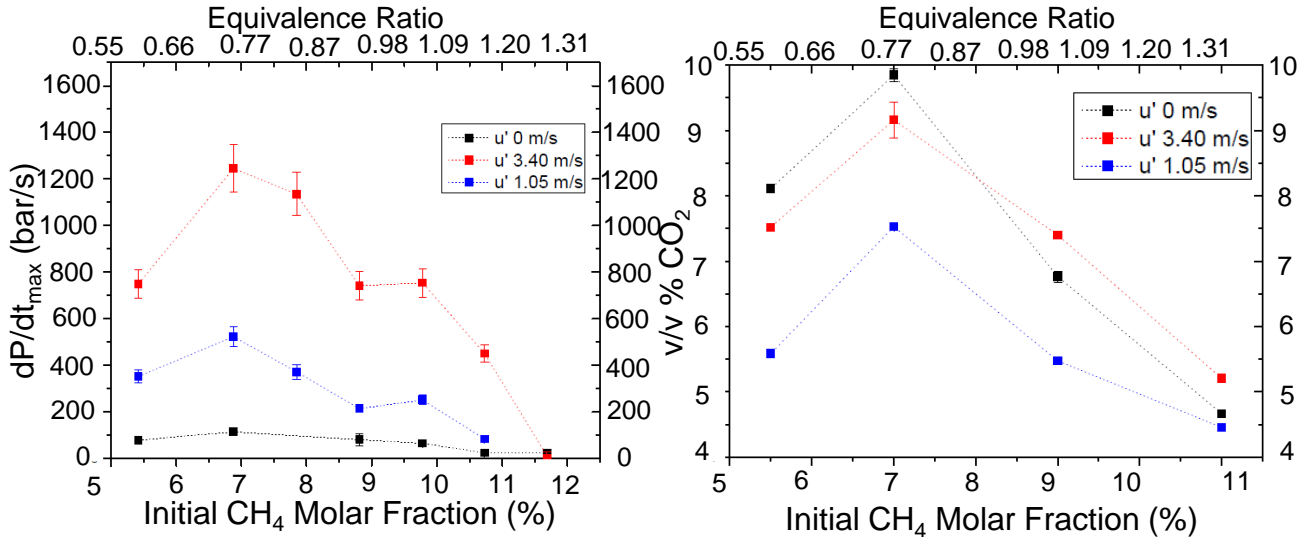
Modifications on the combustion reaction could be related to:

- ① Radiation heat exchange
- ② Flame surface changes

# INFLUENCE OF CB CONCENTRATION



*Printex XE2/Methane/Air mixture under quiescent conditions.*



*Printex XE2/Methane/Air mixture 2.5 g/m<sup>3</sup>.*

The **highest (dP/dt)<sub>max</sub>** is obtained

↳ Maximum conversion of **CO to CO<sub>2</sub>**

Variables that promotes the conversion of CO<sub>2</sub>

- ① Carbon black addition for **Fuel lean mixtures**
- ② Increase on the initial turbulence level

Complex interactions between the turbulence and the combustion reaction are evidenced

↳ **Test on inert powder**

# CONCLUSIONS

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- Dust clouds of the carbon particles studied remain highly stable after the dispersion
  - Mean mobility diameter
    - Printex XE2 ~ 240nm
    - Corax N550 ~ 370nm
  - **Agglomeration phenomena** is limited because of the low concentrations of powders
- **Explosion severity** of methane/air mixtures increases when low concentrations of carbon black nanoparticles are added
  - Even at **low initial turbulence**
- The **specific surface area** of particles have a great influence on the explosion severity
  - **Deagglomeration phenomena** at high turbulence systems

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